## A Long-range Intramolecular Functionalization by Alkoxyl Radicals: a Long-range Intramolecular Oxygenation of C(15) of the Androstane Skeleton<sup>1</sup>

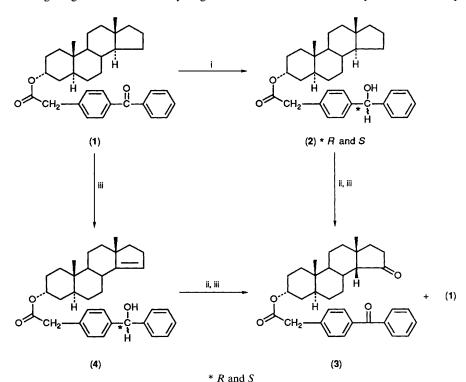
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The first example of a one-step introduction of a carbonyl group to C(15) of the  $5\alpha$ -androstane skeleton, based on a long-range intramolecular hydrogen abstraction by alkoxyl radicals, generated by irradiation of  $5\alpha$ -androstane esters carrying a benzhydryl group in carbon tetrachloride containing mercury(II) oxide and iodine, is described.

In an earlier paper,<sup>2</sup> we reported a two-step long-range intramolecular hydroxylation of C(25) of the cholestane side-chain, based on a long-range intramolecular hydrogen

abstraction by alkoxyl radicals generated by the irradiation of hypoiodites of esters carrying a benzhydryl group. In this communication we report the one-step introduction of a

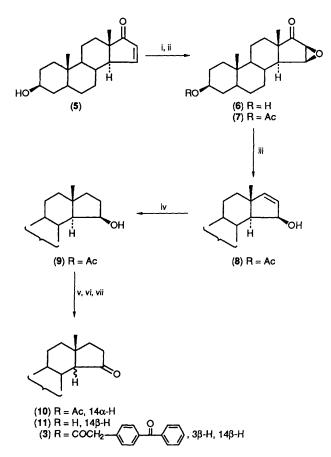


Scheme 1. Reagents and conditions: i, NaBH<sub>4</sub>-MeOH, 25 °C; ii, HgO-I<sub>2</sub>-CCl<sub>4</sub>; iii, hv.

carbonyl group to C(15) of the androstane skeleton, based on the same strategy for intramolecular hydroxylation.<sup>2</sup> Thus,  $5\alpha$ -androstan- $3\alpha$ -yl-4-(hydroxyphenylmethyl)phenyl acetate (2)<sup>†</sup> was prepared by reducing the corresponding esters (1),<sup>†</sup> derived from benzophenone-4-carboxylic acid and  $5\alpha$ -androstan- $3\beta$ -ol, with NaBH<sub>4</sub> (Scheme 1). Compound (2) was a mixture of epimers of the carbon carrying the hydroxy group. Irradiation of the hypoiodite of (2) in carbon tetrachloride [prepared *in situ* with three equivalents of mercury(II) oxide and iodine] with a 450-W high-pressure Hg arc in a nitrogen atmosphere gave a mixture of products from which the 15-ketone (3)<sup>†</sup> arising from a long-range hydrogen abstraction and the parent ester (1) were isolated in 20 and 51% yields<sup>‡</sup> by means of preparative TLC.

The structure of ketone (3) was established as  $15 - 0x_0 - 5\alpha$ ,-14 $\beta$ -androstan-3 $\alpha$ -yl(4-benzoyl)phenylacetate by means of IR, MS, and <sup>1</sup>H NMR spectroscopy as well as by its independent synthesis, as outlined in Scheme 2. 17-Oxo-5 $\alpha$ and rost-15-en-3 $\beta$ -ol (5),<sup>3</sup> obtained from epiandrosterone, was transformed into  $15\beta$ ,  $16\beta$ -epoxide (6)<sup>†</sup> with 30% hydrogen peroxide and sodium hydroxide (4 M) at room temperature. The epoxide (6) was converted to acetate (7),  $\dagger$  which gave  $3\beta$ -acetoxy- $5\alpha$ -androst-16-en- $15\beta$ -ol (8)<sup>†</sup> in 35% yield by treatment with 90% hydrazine and glacial acetic acid in methanol according to the procedure of Wharton.<sup>4</sup> Catalytic hydrogenation of allylic alcohol (8) in ethanol over 5% Pd/C gave  $3\beta$ ,  $15\beta$ -dihydroxy- $5\alpha$ -androstane 3-acetate (9)† in 75% yield. Oxidation of  $15\beta$ -ol (9) in dichloromethane with pyridinium chlorochromate (PCC) gave  $3\beta$ -acetoxy- $5\alpha$ androstan-15-one (10)<sup>†</sup> in 99% yield. Heating a solution of ketone (10) in benzene containing 5% KOH-methanol under reflux gave  $15 - 0x_0 - 5\alpha$ ,  $14\beta$ -androstan- $3\beta$ -ol (11)<sup>†</sup> in 73%

<sup>‡</sup> The yields are based on the converted starting materials.



Scheme 2. Reagents and conditions: i, 30% H<sub>2</sub>O<sub>2</sub>-NaOH (4 M)-Bu'OH, 17 h; ii, Ac<sub>2</sub>O-pyridine, 25 °C; iii, 90% NH<sub>2</sub>NH<sub>2</sub>-MeOH-AcOH, room temp.; iv, Pd/C-EtOH-H<sub>2</sub>, room temp.; v, PCC-CH<sub>2</sub>Cl<sub>2</sub>, room temp.; vi, 5% KOH-MeOH, reflux, 1 h; vii, PhCOC<sub>6</sub>H<sub>4</sub>CH<sub>2</sub>CO<sub>2</sub>H-PPh<sub>3</sub>-EtO<sub>2</sub>CN=NCO<sub>2</sub>Et-THF, room temp., 20 h.

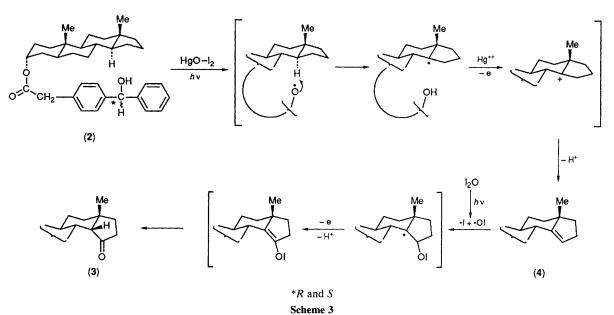
yield. Its reaction with 4-benzoylphenyl acetic acid in the presence of triphenylphosphine and diethylazodicarboxylate<sup>5</sup> in dry tetrahydrofuran (THF) gave 15-oxo- $5\alpha$ ,14 $\beta$ -androst-an- $3\alpha$ -yl 4-benzoylphenyl acetate as a glass, identical with ketone (3) obtained in the long-range reaction (*vide supra*).

The experiments described below proved that ketone (3) is formed via alkene (4)<sup>†</sup> as an intermediate. Thus, the irradiation of androstane alkene (4)<sup>6</sup> in carbon tetrachloride containing mercury(II) oxide and iodine under the same conditions as for ester (2) gave *D*-ring ketone (3) in 32% yield. The pathway for formation of (3) is outlined in Scheme 3. Thus, the alkoxyl radical generated from esters (2) abstracts the C(14) hydrogen to give alkene (4) first. The addition of the iodoxyl radical to (4), followed by a one-electron oxidation of the resulting radical, gives the *D*-ring ketone (3).

The long-range intramolecular functionalization reported herein is the first example in which a carbonyl group is introduced to a remote position in one step as the result of a long-range intramolecular hydrogen abstraction involving a 1,17-hydrogen transfer.§

<sup>†</sup> Selected data for (2): a glass;  $v_{max.}$  (neat) 3450 (OH), 1740 (C=O), and 1250 cm<sup>-1</sup> (C–O);  $\delta_{H}$  (270 MHz) 0.69 (3H, s, 19-H), 0.77 (3H, s, 18-H), 3.59 (2H, s, COCH<sub>2</sub>), 5.00 (1H, quint., J 2.6 Hz, 3β-H), and 5.83 (1H, d, J 3.3 Hz, CHOH); m/z (FD MS used throughout) 500  $(M^+, 100\%)$  and 483 [ $(M - OH)^+, 93$ ]. For (1): m.p. 135–137 °C (light petroleum):  $v_{max}$  (Nujol) 1735 (C=O) and 1655 cm<sup>-1</sup> (PhC=O);  $\delta_{\rm H}$  (270 MHz) 0.68 (3H, s, 19-H), 0.77 (3H, s, 18-H), 3.71 (2H, s, COCH<sub>2</sub>), and 5.04 (1H, quint., J 2.6 Hz, 3β-H); m/z 498 (M<sup>+</sup>, 100%). For (3): a glass;  $v_{max}$ . (Nujol) 1725 (C=O) and 1655 cm<sup>-1</sup> (PhC=O);  $\delta_{H}$  (270 MHz) 0.73 (3H, s, 19-H), 1.15 (3H, s, 18-H), 3.71 (2H, s, COCH<sub>2</sub>), and 5.04 (1H, quint.,  $J 2.6 \text{ Hz}, 3\beta$ -H);  $m/z 512 (M^+, 100\%)$ . For (6): an oil; δ<sub>H</sub> (90 MHz) 0.86 (3H, s, 19-H), 1.15 (3H, s, 18-H), 3.27 (1H, d, J 2.6 Hz, 15-H), 3.57 (1H, m, 3-H), and 3.81 (1H, d, J 2.6 Hz, 16-H). For (7): an oil;  $v_{max}$  (neat) 1730 (C=O) and 1248 cm<sup>-1</sup> (C-O);  $\delta_{H}$  (90 MHz) 0.88 (3H, s, 19-H), 1.15 (3H, s, 18-H), 2.02 (3H, s, OCOMe), 3.27 (1H, d, J 2.6 Hz, 15-H), and 3.81 (1H, d, J 2.6 Hz, 16-H); m/z 346 (M<sup>+</sup>, 100%), 330 [(M-16)<sup>+</sup>, 6], and 287 [(M-OCOMe)<sup>+</sup>, 16%]. For (8): m.p. 157–159 °C (hexane); v<sub>max</sub>. (Nujol) 3458 (OH) and 1714 cm<sup>-1</sup> (C=O);  $\delta_{\rm H}$  (90 MHz) 0.90 (3H, s, 19-H), 1.83 (3H, s, 18-H), 2.02 (3H, s, OCOMe), 4.53 (1H, br. s, 15α-H), 4.70 (1H, br. s, 3α-H), 5.86 (1H, dd, J 5.8 and 2.6 Hz, 16-H), and 6.20 (1H, d, J 5.8 Hz, 17-H). For (9): m.p. 118.5-119.5 °C (hexane);  $v_{max}$  (Nujol) 3514 (OH) and 718 cm<sup>-1</sup> (C=O);  $\delta_{H}$  (90 MHz) 0.86 (3H, s, 19-H), 0.99 (3H, s, 18-H), 2.01 (3H, s, OCOMe), 4.30 (1H, br. s, 15 $\alpha$ -H), and 4.70 (1H, br. s, 3 $\alpha$ -H). For (10): m.p. 149--154 °C (light petroleum);  $v_{max}$ . (Nujol) 1730 (C=O) and 1244 cm<sup>-1</sup> (C-O);  $\delta_{\rm H}$  (90 MHz) 0.79 (3H, s, 18-H), 0.83 (3H, s, 19-H), 2.01 (3H, s, OCOMe), and 4.68 (1H, br. s, 3 $\alpha$ -H). For (11): m.p. 166—167.5 °C (hexane-acetone);  $v_{max.}$  (Nujol) 3280, 3178 (OH), and 1739 cm<sup>-1</sup> (C=O);  $\delta_{H}$  (90 MHz) 0.76 (3H, s, 19-H), 1.15 (3H, s, 18-H), 2.0-2.8 (3H, m, 14- and 16-H), and 3.59 (1H, br. s, 3α-H). For (4): a glass;  $v_{max}$  (neat) 1732 (C=O) and 1662 cm<sup>-1</sup> (PhC=O);  $\delta_{H}$  (270 MHz) 0.73 (3H, s, 19-H), 1.15 (3H, s, 18-H), 3.72 (2H, s, COCH<sub>2</sub>), and 5.03 (1H, br. s, 3β-H).

<sup>§</sup> Satisfactory analytical and spectral results were obtained for all the compounds described in this paper.



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